Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

Shao-Wei Peng, Yan-Li Miao and Wen-Dong Song*

College of Science, Guang Dong Ocean University, Zhan Jiang 524088, People's Republic of China

Correspondence e-mail: songwd60@126.com

Key indicators

Single-crystal X-ray study T = 293 K Mean σ (C–C) = 0.002 Å R factor = 0.025 wR factor = 0.069 Data-to-parameter ratio = 13.8

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

Diaquabis[5-(pyrazin-2-yl)tetrazolato]manganese(II)

In the title complex, $[Mn(C_5H_3N_3)_2(H_2O)_2]$, the Mn^{II} atom, located on an inversion center, is coordinated by four N atoms from two 5-(2-pyrazinyl)tetrazolate ligands and two water molecules in a distorted octahedral geometry. The packing is governed by intermolecular $O-H\cdots N$ hydrogen bonds.

Received 26 November 2006 Accepted 12 December 2006

Comment

The design, synthesis, characterization, and properties of supramolecular networks formed by using functionalized organic molecules as bridges between metal centers are of great interest (Rizk *et al.*, 2005; Eddaoudi *et al.*, 2001). Reports on tetrazoles are increasing rapidly, since tetrazoles play an important role as a ligand in coordination chemistry (Song & Xi, 2006). Recently, 5-substituted 1*H*-tetrazoles have been synthesized by a facile approach (Demko & Sharpless, 2001*a,b*). In the general reaction, the tetrazoles are prepared by the addition of an azide to a nitrile in water with the aid of a Lewis acid such as Zn^{2+} . In this paper, we selected 2-cyanopyrazine, NaN₃ and a Lewis acid MnCl₂ as reagents to yield, in one-step, the title mononuclear complex, (I), under hydro-thermal conditions.



In (I), the Mn^{II} atom, located on an inversion center, is coordinated by four N atoms from two 5-(2-pyrazinyl)tetrazolate ligands and two water molecules in a distorted octahedral geometry. In each ligand, the pyrazinyl and tetrazolyl rings are almost coplanar, making a dihedral angle of 2.51 (5)°. Intermolecular $O-H \cdots N$ hydrogen bonds (Table 2) stabilize the crystal packing.

Experimental

© 2007 International Union of Crystallography All rights reserved Hydrothermal treatment of $MnCl_2 \cdot 4H_2O$ (1.0 mmol, 0.197 g), 2-cyanopyrazine (1 mmol, 0.105 g), NaN_3 (1 mmol, 0.065 g), and water

metal-organic papers

(3 ml) over 50 h at 422 K yielded red block-shaped crystals (yield: 60%).

Z = 2

 $D_{\rm r} = 1.733 {\rm Mg} {\rm m}^{-3}$

 $0.25 \times 0.20 \times 0.17 \text{ mm}$

9055 measured reflections 1700 independent reflections

1576 reflections with $I > 2\sigma(I)$

 $w = 1/[\sigma^2(F_0^2) + (0.0389P)^2]$

where $P = (F_0^2 + 2F_c^2)/3$

+ 0.1934P]

 $(\Delta/\sigma)_{\rm max} < 0.001$ $\Delta \rho_{\rm max} = 0.17 \text{ e} \text{ Å}^{-3}$

 $\Delta \rho_{\rm min} = -0.28 \text{ e} \text{ Å}^{-3}$

Mo Ka radiation

 $\mu = 0.93 \text{ mm}^{-1}$

T = 293 (2) K

Block, yellow

 $R_{\rm int} = 0.022$

 $\theta_{\rm max} = 27.5$

Crystal data

 $\begin{bmatrix} Mn(C_5H_3N_3)_2(H_2O)_2 \end{bmatrix} \\ M_r = 385.24 \\ Monoclinic, P2_1/n \\ a = 5.9840 (3) Å \\ b = 11.7421 (5) Å \\ c = 10.8646 (5) Å \\ \beta = 104.731 (2)^{\circ} \\ V = 738.31 (6) Å^3 \end{bmatrix}$

Data collection

Bruker APEX-II area-detector diffractometer φ and ω scans Absorption correction: multi-scan (*SADABS*; Sheldrick, 1996) $T_{\min} = 0.800, T_{\max} = 0.858$

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.025$ $wR(F^2) = 0.069$ S = 1.051700 reflections 123 parameters H atoms treated by a mixture of independent and constrained refinement

 Table 1

 Hydrogen-bond geometry (Å, °).

$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - H \cdot \cdot \cdot A$
$ \begin{array}{c} O1W - H1W \cdots N6^{i} \\ O1W - H2W \cdots N4^{ii} \end{array} $	0.83 (2)	1.92 (2)	2.7425 (16)	169 (2)
	0.77 (3)	2.04 (3)	2.7939 (18)	167 (2)

Symmetry codes: (i) $-x + \frac{3}{2}$, $y + \frac{1}{2}$, $-z + \frac{1}{2}$; (ii) x + 1, y, z.

H atoms of the water molecule were located in a difference Fourier map and were refined isotropically. Other H atoms were placed in calculated positions and treated as riding on their parent atoms, with C-H = 0.93 Å and $U_{iso}(H) = 1.2U_{eq}(C)$.



Figure 1

The molecular structure of (I), showing the atom-numbering scheme. Non-H atoms are shown as 50% probability displacement ellipsoids. H atoms are represented as small spheres of arbitrary radii. [Symmetry code: (i) 2 - x, 2 - y, 1 - z.]

Data collection: *SMART* (Bruker,1998); cell refinement: *SAINT* (Bruker, 1999); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker, 1998); software used to prepare material for publication: *SHELXTL*.

The authors acknowledge Guang Dong Ocean University for supporting this work.

References

- Bruker (1998). SMART (Version 5.0) and SHELXTL (Version 6.12). Bruker AXS Inc, Madison, Wisconsin, USA.
- Bruker (1999). SAINT. Bruker AXS Inc., Madison, Wisconsis, USA.
- Demko, Z. P. & Sharpless, K. B. (2001a). J. Org. Chem. 66, 7945-7950.

Demko, Z. P. & Sharpless, K. B. (2001b). Org. Lett. 3, 4091-4094.

- Eddaoudi, M., Moler, D. B., Li, H. L., Chen, B. L., Reineke, T. M., O'Keeffe, M. & Yaghi, O. M. (2001). Acc. Chem. Res. 34, 319–330.
- Rizk, A. T., Kilner, C. A. & Halcrow, M. A. (2005). CrystEngComm, 7, 359-362.
- Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.

Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.

Song, W.-D. & Xi, D.-L. (2006). Acta Cryst. E62, m2841-m2842.